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The localised density of H₂O₂ in the effluent of a cold atmospheric pressure plasma jet determined by continuous-wave cavity ring-down spectroscopy

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Although the research on cold atmospheric pressure plasma jets and their applications is steadily growing, several questions remain open regarding fundamental aspects of how reactive species, such as hydrogen peroxide (H_2O_2) , are generated in cold atmospheric pressure plasma jets, and how the composition of reactive species can be tailored for a specific purpose. Accordingly, absolute and spatially resolved distributions of the densities of reactive species in the effluent of cold atmospheric pressure plasma jets are required. In this work, a time efficient way to determine the local distribution of gas phase H₂O₂ in the effluent of a cold atmospheric-pressure plasma jet using continuous-wave cavity ring-down spectroscopy at a wavelength of 8.12 μ m is presented. By a combination of an axial scan and of several radial distributions, the localised density distribution of H_2O_2 in the effluent of the kINPen-sci plasma jet was obtained. Therefore, the effective absorption length was determined from the evolution of the radial distributions as a function of the distance from the nozzle, which was 1.6 mm close to the nozzle of the plasma jet, and increased to approximately 5 mm at a distance of 10 mm from the nozzle. The maximum density of approximately 2 · 10¹⁴ cm⁻³ was found in the centre of the effluent close to the nozzle. From the presented localised density distribution, it can be concluded that H_2O_2 is significantly generated within the plasma zone of the plasma jet. This work presents an important step towards the understanding of formation and consumption mechanisms of biomedically relevant species in the plasma zone and the effluent of a cold atmospheric pressure plasma jet.

KEYWORDS

cold atmospheric pressure plasma jet, Abel inversion, spatial distribution, cavity ring-down spectroscopy, hydrogen peroxide

1 Introduction

With the rising number of applications for cold atmospheric pressure plasma jets (CAPJs), the demand of sophisticated diagnostic techniques also increases. A challenge for the diagnostics of CAPJs is the small diameter over which reactive species are distributed [1-5]. Most of the standard diagnostic techniques are well established for low pressure plasmas in large chambers and have to be strongly modified to be used for CAPJs [6]. This is cumbersome, since it requires the entire adaptation of the measurements methodology and data analysis. Moreover, some of the species are often present only in trace amounts,

although they are important for the application. Hence, a high sensitivity for the detection of species is required, which can be obtained, for instance, by the employment of optical cavities [7-11].

One of the key species for biomedical applications is hydrogen peroxide (H₂O₂), as it has been proven to be beneficial for both, cell growth and cell death [12-18]. As a signalling agent, H₂O₂ is involved in several reactions occurring in cells and leads at high concentration also to cell inhibition [19-22]. With the development of cold atmospheric pressure plasma jets (CAPJs), a non-thermal plasma source for several reactive hydrogen, oxygen and nitrogen species including H₂O₂ has been provided, which operates at atmospheric pressure with gas temperatures remaining around room temperature and which is suitable for localised treatments due to its small dimensions; commonly, the volume, in which the reactive species are distributed, is in the order of mm [1-5]. In particular, the fields of plasma medicine, materials processing of heat sensitive targets, and of plasma agriculture have been evolving strongly due to the employment of CAPJs. The number of applications for CAPJs is continuously rising, such that the adaptability of the composition of reactive species to a specific purpose gains importance. Therefore, a thorough understanding of the chemical reactions occurring in the plasma zone and the effluent is crucial, which requires spatially resolved density distributions of several reactive species envolved into the reaction network.

H₂O₂ is mainly generated by the reaction of two hydroxyl radicals (OH); either in the gas phase or in a liquid. Regarding CAPJs, H₂O₂ was mainly investigated within a liquid [22-28]. For this, often test stripes or colorimetric assays have been employed. Only a few investigations of detecting H₂O₂ in the gas phase with absorption spectroscopy methods have been reported [19, 29]. Basically, there are two absorption features for H₂O₂ that have a sufficiently high absorption cross section suitable for absorption spectroscopy: The v_6 -band of the asymmetric OH-bending between 1,175 and 1,340 cm⁻¹ (8.510–7.460 μ m) with a maximum line strength of approximately $3.9 \cdot 10^{-20} \text{ cm}^2 \cdot \text{cm}^{-1}$ [30-32], and the v_5 -band of the asymmetric OH-stretching at approximately 3,600 cm⁻¹ (2.778 μ m) with a line strength of approximately $1.2~\cdot~10^{-20}\,cm^2 \cdot cm^{-1}$ [30, 33]. However, in particular with the operation of CAPJs in the open air, the measured absorption is a superposition of absorption features of H2O2 that overlap significantly with absorption features of water (H₂O). Hence, the spectral region to detect H₂O₂ has to be chosen carefully. Winter et al. and Schmidt-Bleker et al. have reported previously the determination of H₂O₂ in the kINPen plasma jet by means of Fourier transform absorption spectroscopy [19, 29]. In these experiments, the gas of the effluent was collected into a large box including a multi pass cell thereby increasing the absorption length to approximately 2 m [19]. Up to now, there is no work published, where H₂O₂ is detected in the effluent of a CAPJ directly.

In this work, it is demonstrated that the density of gas phase H_2O_2 can be determined by continuous-wave cavity ring-down spectroscopy (cw-CRDS) directly in the effluent of a CAPJ at a wavelength of approximately 8 μ m. For the determination of absolute densities, the effective absorption length is crucial, in particular for CAPJs, as this is often unknown. Henceforth, in this work, radial scans have been performed in order to determine the effective absorption lengths for H_2O_2 in the

effluent. Moreover, spatially resolved information was obtained by applying an Abel inversion on the radial scans at various axial positions, and by a subsequent interpolation between the determined radial density distributions at various axial positions. All the measurements presented have been performed on the kINPen-sci plasma jet.

The remainder of this paper is structured as follows: Firstly, in Section 2, the experimental setup for the determination of H₂O₂ in the effluent of the kINPen-sci plasma jet by means of cw-CRDS is presented. Subsequently, in Section 3, the procedure for the data analysis is explained. After an illustration of the relevant spectroscopic parameters for H₂O₂, the fitting procedure for full spectra and the on/off-resonance method used for the determination of the localised distribution for H₂O₂ are elucidated. In Section 4, the determined line-of-sight integrated densities and pressure broadening coefficients for H2O2, which were obtained from full spectra, are discussed. By performing radial scans, the effective absorption lengths were determined, which are presented together with the axial and radial density distributions of H₂O₂. An Abel inversion was applied on the radial scans in order to obtain spatially resolved densities from line-of-sight integrated measurements. Based on the radial and axial distributions, the localised distribution for H2O2 in the effluent of the kINPen-sci is presented. Finally in Section 5, the conclusions of this work are drawn.

2 Experimental setup

The kINPen-sci plasma jet, a cold atmospheric pressure plasma jet with a needle-to-grounded-ring-electrode configuration in a dielectric capillary with a diameter of 1.6 mm, which was investigated in this work, was operated with 3 slm argon (Ar) at a frequency of 860 kHz [34]. By employing a gas curtain of 5 slm oxygen (O_2) , the surrounding atmosphere was controlled [35]. 10% of the Ar feed gas were guided through a bubbler with destilled water at room temperature in order to obtain a feed gas humidity of approximately 1,600 ppm. The feed gas humidity was measured with a dew point hygrometer and by cavity ring-down spectroscopy. For the latter, the cavity presented in Figure 1 was closed except for a small exit hole in the centre of the cavity ($\approx 1 \text{ mm}^2$), and filled with the humid Ar gas. The water concentration was obtained from a spectrum recorded between 1,233.04 and 1,233.41 cm⁻¹. In order to vary the position, the kINPen-sci plasma jet was mounted on an xyzstage, where the z-axis was defined as the symmetry axis of the plasma jet, and the y-axis was parallel to the propagation of the laser beam within the optical cavity used for cavity ring-down spectroscopy.

In Figure 1, a schematic of the continuous-wave cavity ringdown spectroscopy setup for the determination of H_2O_2 is depicted, which is similar to the setup described by [36]. The cavity was composed of two high reflective mirrors (Lohnstar Optics, reflectivity: 99.98%) separated by a distance of 54.5 cm, while the first cavity mirror was positioned on a piezo-electric ring actuator (RA12-24, Piezosystem Jena, total stroke: 14.8 μ m). With a triangular function generated by a frequency generator (AFG 3000 C, Tektronix, frequency: 100 Hz) applied to the piezodriver, the cavity length was constantly varied by 4 μ m in order



to increase the coupling efficiency of the laser to the cavity. To protect the cavity from dust particles, two metallic tubes with a diameter of 5 cm and a length of approximately 20 cm each were mounted on the holders of the cavity mirrors. The tubes were purged by 5 slm of N₂ with a gas inlet close to the mirrors to reduce the amount of dust and of water in the beam path. As a laser source, a quantum cascade laser in an HHL-package [HHL-223, Alpes Lasers, tuning range: 1,224-1,234 cm⁻¹ (8.170-8.106 µm)] was employed, which was operated with a low-noise QCL current driver (QCL1000, Wavelength Electronics) and a chassis mount temperature controller (PTC5K-CH, Wavelength Electronics). The beam of the QCL was directed through an acousto-optic modulator (AOM) (1208-G80-4, Isomet, frequency: 80 MHz), while the 0th diffraction order was guided by four gold coated mirrors and two apertures to a wavelength analyser (Laser Spectrum Analyzer 771 B, Bristol Instruments, $1-12 \mu m$). The 1st diffraction order (85% of the input laser power) was guided by two other gold coated mirrors, the coupling mirrors, to the optical cavity. In order to match the beam shape of the laser to the cavity modes, two mode matching lenses (plano-convex lens with ZnSe substrate, Thorlabs, focal length: 75 mm) were positioned at 4 cm behind the first coupling mirror M1 in the case of L1, and at 13 cm in front of the second coupling mirror M2 in the case of L2 as illustrated in Figure 1. The transmitted laser beam after the cavity was focused by an off-axis parabolic mirror onto a fast detector (PVI-4TE-8-1x1, Vigo systems, 790 MHz high cut-off frequency). With an oscilloscope (Waverunner Xi-A, Teledyne LeCroy, band width: 400 MHz, sample rate: 5 GS/s) the detector signal was recorded. The oscilloscope was triggered, when the detector signal reached a threshold value preset at a delay generator (DG535, Stanford Research Systems, preset threshold value: 1 V) that was also connected to the detector output. When the detector signal reached the threshold value set at the delay generator, the AOM was switched off in order to initiate the observation of a ring-down event. The data from the oscilloscope were transferred to a personal computer (PC) and analysed by a custom LabView program. With the same LabView program, the laser current and temperature were set and measured at the same time by using a Bayonet Neill-Concelman connector board (BNC-board, National instruments) connected to the same PC.

In order to determine the stability of the cavity ring-down spectrometer, which defines an appropriate number of averages to be taken for the analysis of the ring-down times, a series of 3,600 ring-down times was recorded. Based on these data, an Allan-Werle deviation analysis was performed. The Allan-Werle deviation is a measure for the number of averages that improve the signal to noise ratio. Firstly introduced in 1966 by Allan to determine the stability of atomic clocks [37], the Allan-Werle deviation became a valid method to determine the type of noise [38] and to determine the theoretical detection limit for absorption coefficients obtained by optical cavities [10, 39–41]. In the case of an Allan-Werle deviation, firstly, the mean \bar{y}_i of subsequent measurements y_k with increasing number of samples *i* is calculated:



FIGURE 2

Absorption spectrum obtained from cavity ring-down spectroscopy between 1,233.04 and 1,233.41 cm⁻¹, while the plasma jet was operated with humdid Ar gas (2.7 slm dry Ar mixed with 300 sccm dry Ar guided through a bubbler held at room temperature). (A) Plasma off (B) Plasma on.

$$\bar{y}_i = \frac{1}{i} \sum_{k=1}^{i} y_k.$$
 (1)

The Allan-Werle deviation ADEV is then defined as:

$$ADEV = \sqrt{\frac{1}{2} \left(\bar{y}_{i+1} - \bar{y}_i \right)^2}.$$
 (2)

In Figures 2A, B, two example plots for the Allan-Werle deviation of the cavity losses $\frac{1}{c\tau}$ are depicted as a function of the time that was needed to measure the included number of samples, for the cases

where the kINPen-sci plasma jet was switched off and on, respectively. The time also includes the computation time for the data transfer and analysis to determine the ring-down times. In contrast to common Allan-Werle deviation plots as reported for example, in reference [10], the Allan-Werle deviation was varying by orders of magnitudes from data point to data point resulting in a broad band with an upper and a lower limit. In both cases, plasma off and plasma on, the upper limit for the Allan-Werle deviation decreased inversely proportional to the square root of the averaging time ($ADEV \propto t^{-1/2}$). According to Barnes et al., this corresponds to

uncorrelated "white" frequency noise [42]. The lower limit decreased inversely proportional to the averaging time (*ADEV* $\propto t^{-1}$), which corresponds, according to Barnes et al., to uncorrelated "white" phase modulations [42].

Based on 100 ring-down events, an Allan-Werle deviation of 4.1 $\cdot 10^{-10} \,\mathrm{cm}^{-1}$ was determined for an averaging time of 27 s in both cases, plasma off and plasma on. Compared to the Allan-Werle deviation of the cavity used for the measurements of hydroperoxyl radicals (HO₂) [10], the cavity used in this work for the determination of H₂O₂ is more noisy by approximately a factor of 10. This could be due to the fact, that the cavity used for the determination of the H2O2 density was operated in the open air and purged with nitrogen in contrast to the jet beeing operated in a box with a lid in the case of the measurements by Gianella et al. [10]. The gas flow of the purge gas was along the direction of the laser beam, and perpendicular to the gas flow through the kINPen-sci plasma jet, which could induce additional turbulences to the turbulent gas flow through the kINPen-sci plasma jet and thus lead to fluctuations of the refractive index in the cavity. Notably, in the case when the plasma was on, the broad band of the Allan-Werle deviation as depicted in Figure 2B started to bend, resulting in a distribution parallel to the time axis. This indicates that the cavity became unstable for averaging times larger than 1,000 s, which is equivalent of averaging more than 1,500 samples. However, the values for the Allan-Werle deviation of 100 ring-down events were the same for both cases, plasma off and plasma on. This indicates that the influence of the plasma on the cavity stability is smaller than the fluctuations induced by the gas flow. A more detailed analysis of the influence of the plasma on the stability of the cavity remains for future investigations. By taking an absorption coefficient of 2.3 · 10^{-19} cm² for the investigated H₂O₂ absorption feature into account, the detection limit due to the cavity stability is $2.4 \cdot 10^{11}$ cm⁻³, when averaged over 100 ring-down events as used in the remainder of the work.

3 Determination of absolute number densities for H_2O_2

Line positions and line strengths for transitions corresponding to H₂O₂ together with calculated pressure broadening coefficients in air can be found, for example, in the HITRAN database [43]. The most prominent absorption features around a wavelength of 8 µm result from transitions in the v_6 band centred at around 1,270 cm⁻¹, the asymmetric bending of the OH groups in H_2O_2 [30–32, 43]. The strongest absorption features for H2O2 can be measured at approximately 1,250 cm⁻¹. However, also broadband absorptions of nitric acid (HNO₃), nitrous acid (HONO) and of dinitrogen pentoxide (N2O5) molecules, and absorptions lines from hydroperoxyl radicals (HO₂), methane (CH₄), and nitrous oxide (N_2O) overlap with the H_2O_2 lines in this region. As all of these species tend either to be generated by the kINPen-sci plasma jet or to be present in the laboratory air [29], it is difficult to distinguish between those contributions. Within the range between 1,230.5 and 1,232.0 cm⁻¹, two prominent absorption features for H₂O₂ can be found, while contributions of other species, namely, HNO₃, HONO, and N₂O₂, result in a straight line of broadband absorption. An example for the different absorption cross sections between 1,230.5 and 1,232.0 cm⁻¹ is depicted in Figure 3, which were calculated at atmospheric pressure from line positions and pressure broadening coefficients taken from the HITRAN database [43], and taken from references [44–46]. To determine the densities of H_2O_2 , in this work, the absorption feature of H_2O_2 within the spectral range between 1,230.8 and 1,231.3 cm⁻¹ was chosen.

3.1 Fitting procedure of full spectra for H₂O₂

To determine the density of H_2O_2 from a measured absorption spectrum, a fitting procedure similar to that reported in references [10, 36] has been employed. The absorption spectrum was obtained by measuring the ring-down times $\tau(\nu)$ and τ_0 at different spectral positions, while the plasma jet was switched on and off, respectively, according to:

$$\frac{1}{c\tau(\nu)} - \frac{1}{c\tau_0} = \alpha(\nu)\frac{d}{L}.$$
(3)

Here, $\alpha(\nu)$ is the absorption coefficient, *L* is the cavity length, *d* is the diameter of the volume, in which the absorbing species are located (i.e., the effective absorption length), and *c* is the speed of light in vacuum. The frequency dependence of τ_0 within the applied tuning range of the QCL was small. Hence, τ_0 was considered to be constant over the recorded spectral range of a spectrum.

The measured absorption $\alpha(\nu) \frac{d}{L}$ is a result of the contributions from the absorption of H₂O₂ distributed with a density [H₂O₂] within a volume with a diameter of $d_{H_2O_2}$, and from (broadband) absorptions of other molecules, which were represented by a straight line with the coefficients b_0 (y-intercept) and b_1 (slope):

$$\alpha(\nu)\frac{d}{L} = b_0 + b_1\nu + \frac{d_{H_2O_2}}{L} [H_2O_2]\sigma_{H_2O_2}(\nu) + \epsilon(\nu).$$
(4)

Here, $\sigma_{H_2O_2}(\nu)$ is the frequency dependent absorption cross section of H₂O₂, and $\epsilon(\nu)$ is the residual of the fitting procedure. The frequency dependent absorption cross section was computed by using a Voigt function $V(\nu - \nu_{0,t}; \gamma_{H_2O_2})$ with different values for the pressure broadening coefficient $\gamma_{H_2O_2}$ according to:

$$\sigma_{H_2O_2}(\nu; \gamma_{H_2O_2}) = \sum_t S_t V(\nu - \nu_{0,t}; \gamma_{H_2O_2}).$$
(5)

Here, $\gamma_{H_2O_2}$ denotes the pressure broadening coefficient for H_2O_2 chosen as a constant value, while the considered transitions *t* of H_2O_2 at the frequencies $\nu_{0,t}$ and the corresponding line strengths S_t were taken from the HITRAN database [30–32, 43]. In the effluent of the kINPen-sci plasma jet, a mixture of H_2O , O_2 , air, and Ar is expected, which results in a significantly different pressure broadening coefficient than the one reported in the HITRAN database. The pressure broadening coefficient was assumed to be the same for all transitions, as it was reported in the HITRAN database for the pressure broadening coefficients in air [30–32, 43], and previously done for the pressure broadening coefficients for HO₂ [10].

For the determination of $[H_2O_2]$, and of $\gamma_{H_2O_2}$, the sum of squares of the residuals $\chi^2 = \sum_{j=1}^{N_{\nu}} \epsilon^2 (\nu_j)$ in Eq. 4 at the measured spectral positions ν_j were minimised for different values for $\gamma_{H_2O_2}$, such that $\frac{\partial \chi^2}{\partial \beta} = 0$, which is equivalent to solve a matrix-equation with



the form $\mathbf{M}\beta = \omega$. Here, $\beta = (b_0, b_1, [H_2O_2])$ includes the fitting parameters, **M** is a 3 × 3-matrix:

$$\mathbf{M} = \begin{bmatrix} N_{\nu} & \sum_{j=1}^{N_{\nu}} \nu_{j} & \sum_{j=1}^{N_{\nu}} \sigma_{H_{2}O_{2}}(\nu_{j}) \\ \sum_{j=1}^{N_{\nu}} \nu_{j} & \sum_{j=1}^{N_{\nu}} \nu_{j}^{2} & \sum_{j=1}^{N_{\nu}} \sigma_{H_{2}O_{2}}(\nu_{j}) \nu_{j} \\ \sum_{j=1}^{N_{\nu}} \sigma_{H_{2}O_{2}}(\nu_{j}) & \sum_{j=1}^{N_{\nu}} \sigma_{H_{2}O_{2}}(\nu_{j}) \nu_{j} & \sum_{j=1}^{N_{\nu}} \sigma_{H_{2}O_{2}}(\nu_{j})^{2} \end{bmatrix}, \quad (6)$$

where N_{ν} is the total number of the measured spectral positions, and ω comprises the experimental data and the calculated frequency dependent absorption cross sections $\sigma_{H_2O_2}(\nu_j)$ with the chosen values for $\gamma_{H_2O_2}$:

$$\omega = \frac{1}{c} \left[\frac{\sum_{j=1}^{N_{\nu}} \tau(\nu)^{-1} - \tau_0^{-1}}{\sum_{j=1}^{N_{\nu}} (\tau^{-1} - \tau_0^{-1}) \nu_j} \right].$$
 (7)

The values for β that yielded the lowest of squares of the residuals have been retained as the best fit parameters. The error for $[H_2O_2]$ and for $\gamma_{H_2O_2}$ was evaluated by using the same F-ratio method as described in reference [10]. As the procedure to obtain a full spectrum at various radial and axial positions is time consuming, in this work, also the on/ off-resonance method has been employed presented by [36] for the measurements of HO₂, in particular to determine the effective absorption length, which will be described in the following.

3.2 On/off-resonance method for H₂O₂

Based on the on/off-resonance method, radial and axial scans through the effluent have been performed by recording

100 ring-down events at $v_{on} = 1,231.07 \text{ cm}^{-1}$ for the onresonance position, and at $v_{off} = 1,230.82 \text{ cm}^{-1}$ for the offresonance position. The broadband absorptions of HNO₃, HONO, and N₂O₅ were assumed to be the same at the positions for the on- and the off-resonance such that $\Delta \alpha$ is defined by:

$$\Delta \alpha \coloneqq \frac{1}{c\tau(\nu_{on})} - \frac{1}{c\tau(\nu_{off})} = \frac{d_{H_2O_2}}{L} [H_2O_2] \Delta \sigma_{H_2O_2}.$$
 (8)

Here, *c* is the speed of light in vacuum, $\tau(\nu_{on})$ and $\tau(\nu_{off})$ are the measured ring-down times at the on-resonance and off-resonance position, respectively, while the plasma jet was switched on, $d_{H_2O_2}$ is the absorption length for H_2O_2 , $[H_2O_2]$ is the density of H_2O_2 located in a volume with a diameter of $d_{H_2O_2}$ along the line-of-sight of the laser beam, and $\Delta\sigma_{H_2O_2} \coloneqq \sigma_{H_2O_2}(\nu_{on}) - \sigma_{H_2O_2}(\nu_{off})$ is the difference of the frequency dependent absorption cross sections at the on-resonance and the off-resonance position. The assumption for a constant value for the broadband absorbers is valid, as long as the slope for the baseline in the fitting procedure is small compared to the error of the measured absorption.

4 Results and discussion

4.1 Pressure broadening coefficients for H₂O₂ obtained from full spectra

Several full spectra at various axial distances z from the nozzle were recorded, in order to validate the on/off-resonance method,



and to determine the value for the pressure broadening coefficient, $\gamma_{H_2O_2}$, which is crucial for the application of the on/off-resonance method. In Figure 4, absorption spectra for various *z*-distances between 3 mm and 10 mm below the nozzle of the kINPen-sci plasma jet are depicted, together with a fit of the baseline and the contribution of the absorption of H₂O₂. For each spectrum, an absorption feature with a shape similar to the H₂O₂ absorption cross sections as reported by the HITRAN database was observed at each *z*-position, while a pressure broadening coefficient was determined

from the best fit as described in Section 3.1. Therefore, 221 transitions for H₂O₂ with line strengths ranging from approximately 10^{-25} cm²·cm⁻¹ to approximately 10^{-20} cm²·cm⁻¹ were considered. In Figure 5, the frequency dependent absorption cross section $\sigma_{H_2O_2}(v; \gamma_{H_2O_2})$ for H₂O₂ at 300 K with a pressure broadening coefficient of $\gamma_{H_2O_2} = 0.06$ cm⁻¹atm⁻¹ is depicted as calculated according to Eq. 5, together with the corresponding line strengths S_t of the considered 221 H₂O₂-transitions, which were taken from the HITRAN database [30–32, 43]. The peak height



FIGURE 5

Frequency dependent absorption cross section for H₂O₂ at 300 K and 1,013 hPa with a pressure broadening coefficient of $\gamma_{H_2O_2} = 0.06 \text{ cm}^{-1} \text{atm}^{-1}$ as calculated according to Eq. 5, together with the corresponding line strengths of the considered H₂O₂ transitions taken from the HITRAN database [43], [30], [31,32].



of the spectra shown in Figure 4 was varying due to a subjacent baseline, whose mean value increased with increasing z-position. This baseline accounts for broadband absorptions resulting from

transitions of HNO₃, HONO, and N₂O₅, which could not be further specified due to their broadband character. In Figure 6, the corresponding pressure broadening coefficients $\gamma_{H_2O_2}$ for the



observed H₂O₂ transitions determined from the spectra shown in Figure 4 are depicted as a function of the distance from the nozzle z. Within the error, no significant relation between the z-position of the kINPen-sci plasma jet, and the pressure broadening coefficient for H2O2 could be observed. This indicates that, within the margin of error, a change of the gas mixture at various z-positions due to the diffusion of the gas curtain into the effluent was not influencing the pressure broadening coefficient of the investigated transitions of H₂O₂. Hence, a mean value of $\gamma_{H_2O_2} = 0.06 \text{ cm}^{-1} \text{atm}^{-1}$ was chosen for all transitions, and for all radial and axial positions. Notably, this value is by a factor of 0.6 smaller than the value for the pressure broadening coefficient in air reported in the HITRAN database [30-32, 43]. A similar relation between the pressure broadening coefficients in air and the one determined for the gas mixture in the kINPen-sci plasma jet has been found previously for HO₂ reported by [10]. For the following analysis, $\Delta \sigma$ was determined by using the mean value of the frequency dependent absorption cross section to be:

$$\Delta\sigma(H_2O_2) = 1.6 \cdot 10^{-19} \text{ cm}^2.$$
(9)

4.2 Influence of broadband absorptions

As it has been shown that broadband absorptions from other species than H_2O_2 play a significant role in the absorption spectra, the absorption of H_2O_2 has been compared qualitatively to the broadband absorptions. Therefore, the evolution of the baseline was correlated to the evolution of the absorption of H_2O_2 , both as a function of *z*. For all spectra, the slope of the baseline was small compared to the error of the measured absorption. Hence,

considering a single wavelength position is sufficient for the following comparison of the absorption of H2O2 and the baseline. In order to facilitate this comparison, which includes information on the shape of the absorption cross section for H₂O₂ as known so far, the absorption for H₂O₂ was evaluated by assuming a constant absorption length of d = 4 mm and a line-ofsight integrated density for H₂O₂ was determined. In Figure 7, the line-of-sight integrated density of H2O2 is depicted as a function of the distance from the nozzle z, together with the value of the baseline at 1,231.07 cm⁻¹. The density of H_2O_2 increased between z = 3 mmand z = 8 mm linearly by approximately a factor of 2.3 to its maximum 9.7 $\cdot~10^{13}\,cm^{-3}$, and decreased to 7.1 $\cdot~10^{13}\,cm^{-3}$ between z = 6 mm and z = 10 mm. Within the error of the density, the baseline had a similar trend; the baseline increased up to z = 8 mm, and decreased between z = 8 mm and z = 10 mm. This indicates that besides H2O2, also the densities of HNO3, HONO, and N₂O₅ increased. However, the effective absorption length needs to be considered to compare the densities at various z-positions. Due to the on/off-resonance method employed in this work, this can only be performed for the density of H_2O_2 .

4.3 Determination of the effective absorption length for H_2O_2

In order to obtain the effective absorption length, radial scans at various *z*-positions between z = 3 mm and z = 10 mm have been recorded by using the on/off-resonance method. Due to the diameter of the observed cavity mode, which was 5.5 mm in the focal point of the cavity (containing 99% of the intensity), no ring-down measurement closer than 3 mm below the nozzle could be



performed. In Figure 8, radial scans for $\Delta \alpha$ as a function of the distance to the symmetry axis through the kINPen-sci plasma jet, *x*, are depicted for *z* = 3, 4, 6, 8, and 10 mm together with a Gaussian fit, and a line representing the effective absorption length. For all *z*-positions, a Gaussian distribution for $\Delta \alpha$ was determined.

In Figure 9, the centre positions, x_c , for the Gaussian fits of the radial scans are illustrated as a function of the distance from the nozzle *z*. Here, $x_c = 0$ mm denotes the position of the symmetry axis

of the kINPen-sci plasma jet in the centre of the nozzle. For the measured *z*-positions, the centre of the Gaussian fits were distributed between -0.4 and 0.4 mm relative to the centre of the nozzle. However, the weighted mean of the centre positions was at -0.12 mm, which indicates a slight asymmetry in the H₂O₂ distribution. A similar asymmetry has been reported previously for HO₂ [36] and for O and H atoms in the same plasma jet [36]. Within the plasma zone of the plasma jet, the discharge is likely to be



asymmetric because of inhomogeneities of the powered needle electrode, for instance, which results in an asymmetric production of reactive species.

The effective absorption length d was determined by the limits that contain 99% of the area A(z) of the Gaussian fit:

$$d(z) = 2.576 \cdot w(z). \tag{10}$$

Here, w(z) is the width of the Gaussian function, which is defined by:

$$\Delta \alpha(x,z) = \frac{A(z)}{w(z)\sqrt{\frac{\pi}{2}}} \exp\left\{-2\left(\frac{x-x_c}{w(z)}\right)^2\right\} + \Delta \alpha_{off}.$$
 (11)

Here, x_c is the centre position of the Gaussian fit, and $\Delta \alpha_{off}$ is an offset value.

Although the closest measurement could be taken at 3 mm from the nozzle, the localised density distribution of H₂O₂ close to the nozzle can be obtained by extrapolating the absorption lengths to z =0 mm from the nozzle. Therefore, the effective absorption lengths determined for H₂O₂ have been compared to the absorption lengths d_{HO_2} determined previously for HO₂ [36], and to the region with the diameters d_{H} , d_{O} , where O and H atoms were distributed within the effluent, respectively, as reported in reference [47]. This is a reasonable approach, since reactive species, including H₂O₂, HO₂, H atoms, and O atoms, generated in the plasma zone and in the effluent. Consequently, H2O2 is expected to be present in the same region as other reactive species that are precursors for H₂O₂. Furthermore, it is known from previous investigations that the gas curtain starts to diffuse into the effluent efficiently from 4 mm below the nozzle [48], [35], which introduces further species besides species that already have been present in the plasma zone, and which can result into a further production of H_2O_2 . Hence, the region, where H_2O_2 is present close to the nozzle before the gas curtain starts to diffuse into the effluent efficiently, can be expected to be the same as for other reactive species generated within the plasma zone. In Figure 10, the determined values for $d_H/2$, $d_O/2$, $d_{HO_2}/2$, and $d_{H_2O_2}/2$ for the species distributions of H, O, HO₂, and H₂O₂ are illustrated, as they are related to the radius of the nozzle of the plasma jet to be half of the absorption lengths, as a function of the distance to the nozzle *z* together with a polynomial approximation described by:

$$d_{HO_2,H_2O_2}/2 = 1.6 - 0.1 \cdot z + 0.33 \cdot z^2 - 0.057 \cdot z^3 + 0.0039 \cdot z^4 -9.4 \cdot 10^{-5} \cdot z^5.$$
(12)

By using this polynomial approximation, information close to the nozzle can be extrapolated, which was not possible to obtain experimentally due to the inference of the plasma jet with the laser beam in the cavity.

4.4 Axial density distribution of H_2O_2 (lineof-sight integrated)

Based on the determined effective absorption length $d_{H_2O_2}$ presented in Eq. 12, line-of-sight integrated densities have been obtained as a function from the distance to the nozzle *z* from full spectra and from an axial scan $\Delta \alpha(z)$ recorded by using the on/offresonance method. In Figure 11, the line-of-sight integrated density distributions of H₂O₂ in the effluent of the kINPen-sci plasma jet are depicted as a function of the distance from the nozzle *z*. Within the error, the densities obtained from both methods are in agreement with each other. Between *z* = 3 mm and *z* = 6 mm, the density of H₂O₂ plateaued at approximately 7.5 $\cdot 10^{13}$ cm⁻³ within the error, and decreased linearly for further distances from the nozzle to approximately 5 $\cdot 10^{13}$ cm⁻³ at *z* = 10 mm.



FIGURE 10

Lengths *d*, over which the species O, H, HO₂, and H₂O₂ were distributed as function of the distance to the nozzle *z*. The values for d_O , d_{H} , d_{HO_2} , and $d_{H_2O_2}$ were obtained from the width of the Gaussian fit of the radial scans through the effluent of the plasma jet for H₂O₂, and for HO₂ [36], and from the measures spatial distributions for H and O atoms [47].



Line-of-sight integrated density of H_2O_2 at x = 0 mm as a function of z, obtained by the on/off-resonance method and from a fit of full spectra considering an effective absorption length $d_{H_2O_2}(z)$ as shown in Eq. 12.

4.5 Radial density distribution of H₂O₂

One drawback of the employment of absorption spectroscopy is, however, that only line-of-sight integrated densities are

obtained. In order to transform the line-of-sight integrated absorption into a radial density distribution, an Abel inversion was employed, since the kINPen-sci plasma jet can be approximated by a cylindrical symmetry. By applying an Abel



inversion on the radial scans $\Delta \alpha(x, z)$ presented in Figure 8 according to:

$$\Delta \alpha(\rho) = -\frac{1}{\pi} \int_{\rho}^{R} \frac{d\left(\Delta \alpha(x,z)\right)}{dx} \frac{dx}{\sqrt{\left(x-x_{c}\right)^{2}-\rho^{2}}},$$
(13)

the radial density distribution was obtained. Here, ρ is the radial dimension described by $\rho = \sqrt{(x - x_c)^2 + y^2}$, x_c is the centre position of the radial distributions yielding the distance to the symmetry axis through the nozzle of the plasma jet, and *R* is the effective absorption length divided by 2. As all radial scans could be represented by a Gaussian function, the analytical solution for the Abel-transformed absorption coefficient $\Delta \alpha(\rho)$ equals:

$$\Delta \alpha(\rho) = \frac{2}{\pi} \frac{A(z)}{w^2(z)} \exp\left\{-2\frac{\rho^2}{w^2(z)}\right\}.$$
 (14)

Here, A(z) and w(z) are the area and the width of the Gaussian distributions for the radial scans, respectively. The radial density distribution $n_{H_2O_2}(\rho)$ of H_2O_2 was then determined by:

$$n_{H_2O_2}(\rho) = \Delta \alpha(\rho) \frac{L}{\Delta \sigma_{H_2O_2}}.$$
 (15)

Here, *L* is the cavity length, and $\Delta \sigma_{H_2O_2}$ is the absorption cross section determined for the on/off-resonance method presented in Eq. 9.

In Figure 12, the radial density distributions for H_2O_2 are illustrated for various *z*-positions. Except for the measurement at z = 4 mm, the maximum intensities decreased with further distance from the nozzle. However, for the measurement at z = 4 mm, a maximum density of twice the value for the maximum density at z = 6 mm and a factor of approximately 1.5 more than the value for the

maximum density at z = 3 mm was obtained. In comparison to the measurements of full spectra and the axial scan along the symmetry axis of the plasma jet at x = 0 recorded by the on/off-resonance method, the value for $\Delta \alpha$ obtained from the radial scan at z = 4 mm is the only value that does not agree with the previous measurements within the error. Furthermore, at z = 4 mm the maximum was shifted by approximately x = 0.4 mm compared to the symmetry axis through the plasma jet, which is opposite to most of the centre positions for the radial scans at other z positions as illustrated in Figure 9. Hence, the absolute values for the radial distribution obtained from the radial scan at z = 4 mm is regarded as an outlier. Notably, the radial distribution for $\mathrm{H_2O_2}$ depends strongly on the data quality for the radial scans. Since the measured absorption is greater by a factor of 3 than the scatter of the data points, larger errors are to be expected, especially when determining the center position.

4.6 Localised density distribution of H₂O₂

The localised density distribution in the radial and axial dimension was determined from an axial and several radial scans measuring $\Delta \alpha$ as a function of *z* and *x*, respectively, by employing the on/off-resonance method. According to Eq. 15, the localised density distribution for H₂O₂ is given by:

$$n_{H_2O_2}(x, y, z) = \frac{L}{w(z)\sqrt{\frac{\pi}{2}}} \frac{\Delta\alpha(x=0, z)}{\Delta\sigma_{H_2O_2}} \exp\left\{-2\frac{\rho^2 - x_c^2}{w^2(z)}\right\}.$$
 (16)

In order to obtain a continuous density distribution, an interpolation for the evolution of the width w(z) determined





from several radial scans as described by Eqs 10, 12, and for $\Delta \alpha(x = 0, z)$ was used. Therefore, $\Delta \alpha(x = 0, z)$ was described by a polynomial approximation:

$$\Delta \alpha_{H_2O_2} \left(x = 0, z \right) = 4.3 \cdot 10^{-8} - 8.9 \cdot 10^{-9} \cdot z + 1.5 \cdot 10^{-9} \cdot z^2 -3.5 \cdot 10^{-9} \cdot z^3 + 3.5 \cdot 10^{-10} \cdot z^4 -1.5 \cdot 10^{-11} \cdot z^5 + 2.3 \cdot 10^{-13} \cdot z^6.$$
(17)

Due to the lack of data points close to the nozzle for $\Delta \alpha$ for H₂O₂ within the first 3 mm below the nozzle, the polynomial approximation in this region was chosen in such a way that the axial density distribution for H₂O₂, which was determined by using the absorption length $d_{H_2O_2}$, was constant within the first 3 mm. This assumption is based on the observation that previously

investigated reactive species are mainly generated within the plasma zone of the plasma jet and their density remains approximately constant over the first 3 mm, as it was determined for HO₂ [36]. In Figure 13, $\Delta\alpha(x = 0, z)$ measured with the on/off-resonance method, is depicted as a function of *z* together with a polynomial approximation for $\Delta\alpha$.

The resulting localised density distribution of H2O2 is illustrated as a contour plot in a plane cut along the symmetry axis through the centre of the nozzle in Figure 14A. Lateral profiles of this contour plot at different z-positions are presented in Figure 14B, respectively. The densities of H₂O₂ were mainly distributed within a cone with a diameter of approximately 1.6 mm at z = 0 mm and 5 mm at z = 10 mm. The maximum density of approximately $2 \cdot 10^{14} \text{ cm}^{-3} \text{ H}_2\text{O}_2$ was obtained in the centre of the effluent between z = 0 mm and z = 4 mm. For larger z-distances, the density of H₂O₂ decreased to approximately 1 · 1014 cm-3 at 6 mm and remained approximately constant thereafter. From the presented localised density distribution, it can be concluded that H₂O₂ is significantly generated within the plasma zone of the plasma jet, as the maximum density was obtained close to the nozzle. The decrease of the density of H_2O_2 between z = 4 mm and z = 6 mm is most likely due to an impact of the surrounding gas composition; a larger amount of O₂ from the gas curtain leads to a net consumption of H₂O₂. A similar impact of the surrounding gas composition has also been reported for O atoms, H atoms [47], and HO₂ radicals [36]. With the strong reduction of species, that consume H₂O₂, such as OH or HO₂ radicals, the density of H₂O₂ is expected to remain constant at distances further from the nozzle, which is in agreement with the measured densities at distances further than 6 mm from the nozzle.

5 Conclusion

In this work, it was successfully demonstrated how localised densities of H2O2 can be obtained in the effluent of a cold atmospheric pressure plasma jet by means of continuous wave cavity ring-down spectroscopy. Based on several full spectra recorded between 1,230.9 and 1,231.3 cm⁻¹ at various distances from the nozzle of the kINPen-sci plasma jet, the pressure broadening coefficient for H_2O_2 in the specific gas mixture of the effluent has been determined to be 0.06 cm⁻¹atm⁻¹. A baseline for the fits was used to account for underlying broadband absorptions of other species, such as HONO, N2O5 and HNO3. From radial scans obtained by the on/off-resonance method, the effective absorption length was determined, which was 1.6 mm close to the nozzle of the plasma jet, and increased to approximately 5 mm at a distance of 10 mm from the nozzle. By applying an Abel inversion on the radial scans, radial density distributions were obtained, which could be represented by a Gaussian distribution. The localised density distribution for H₂O₂ was determined by a combination of an axial scan along the symmetry axis of the plasma jet and the evolution of the width of the radial distributions. A maximum density of approximately $2 \cdot 10^{14} \, \mathrm{cm^{-3}}$ was found in the centre of the effluent close to the nozzle of the plasma jet and up to z = 4 mm. With increasing z-distance further than 4 mm, the density decreased to $1 \cdot 10^{14}$ cm⁻³ at 6 mm and remained approximately constant thereafter. Regarding the presented localised density distribution for H₂O₂, it can be concluded that H₂O₂ was mainly produced within the plasma zone of the plasma jet, and partially consumed by the impact of the gas curtain between 4 and 6 mm below the nozzle. All in all, the demonstration to obtain the localised density of H₂O₂ in the effluent of a cold atmospheric pressure plasma jet represents an important step towards the investigation of the formation and consumption mechanisms of biomedically relevant species in the effluent of cold atmospheric pressure plasma jets.

Data availability statement

The datasets presented in this study can be found in online repositories. The names of the repository/repositories and accession number(s) can be found below: Name of the repository: INPTDAT; Link to the datasets: https://doi.org/10.34711/inptdat.727.

Author contributions

JH and S-JK contributed to the conception and the design of the study. S-JK and LK performed the data acquisition and the data analysis procedure. S-JK wrote the first draft of the manuscript. All authors contributed to the article and approved the submitted version.

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Conflict of interest

The authors declare that the research was conducted in the absence of any commercial or financial relationships that could be construed as a potential conflict of interest.

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